

Supporting Information

Intramolecular Aryl Migration from Tin to Carbon via a Radical Atom-Transfer Process (ol005871y)

Katsuyu Wakabayashi, Hideki Yorimitsu, Hiroshi Shinokubo, and Koichiro Oshima*

*Department of Material Chemistry, Graduate School of Engineering, Kyoto University,
Kyoto 606-8501, Japan*

Instrumentation and Materials

^1H NMR (300 MHz) and ^{13}C NMR (75.3 MHz) spectra were taken on a Varian GEMINI 300 spectrometer in CDCl_3 as a solvent, and chemical shifts were given in δ value with tetramethylsilane as an internal standard. IR spectra were determined on a JASCO IR-810 spectrometer. TLC analyses were performed on commercial glass plates bearing 0.25 mm layer of Merk Silica gel 60F₂₅₄. Silica gel (Wakogel 200 mesh) was used for column chromatography. The analyses were carried out at the Elemental Analysis Center of Kyoto University. Dichloromethane was dried with molecular sieves 4A. Tetrahydrofuran (THF) was freshly distilled from sodium benzophenone ketyl before use. Benzene was dried over slices of sodium. Grignard reagents were prepared from the corresponding aryl halides and Mg turning (Nacalai tesque, INC). Unless otherwise noted, materials obtained from commercial suppliers were used without further purification.

Experimental Section

Preparation of Aryl-(3-butenyl)dimethyltin

Phenyl magnesium bromide (1.0 M THF solution 110 mL, 110 mmol) was added to a solution of dichlorodimethyltin (11.0 g, 50 mmol) in THF (50 mL) at 0 °C under argon atmosphere. After stirring for 1 h at 0 °C, the reaction mixture was poured into 1 M NH_4Cl solution and was extracted with ethyl acetate. The organic layers were washed with brine, dried and concentrated in vacuo to give almost pure dimethyldiphenyltin (15.1 g, 50 mmol) containing a small amount of

biphenyl in quantitative yield. The product was used for the next reaction without further purification.

Under argon atmosphere, trifluoroacetic acid (3.85 mL, 50 mmol) was added to the solution of dimethyldiphenyltin (15.1 g, 50 mmol) in 50 mL of THF at 0 °C. The resulting mixture was heated at 50 °C with stirring for 4 h. Evaporation of the solvent in vacuo afforded phenyldimethyltin trifluoroacetate as crystalline solid. Quantitative formation of phenyldimethyltin trifluoroacetate was checked by ¹H NMR. The crystal was dissolved in THF (70 mL) and 3-butenylmagnesium bromide (1 M ether solution, 60 mL, 60 mmol) was added at 0 °C under argon atmosphere. After stirring for 2 h, extractive workup followed by silica gel column purification provided 3-butenyldimethylphenyltin (12.6 g, 45 mmol) in 90% yield.

To synthesize functionalized aryltin compounds, further transformations were conducted as follows. Trifluoroacetic acid (0.46 mL, 6.0 mmol) was added to 3-butenyldimethylphenyltin (1.40 g, 5 mmol) in 5 mL of THF at 0 °C. The resulting solution was heated at 50 °C for 4 h with stirring. The solvent and remaining trifluoroacetic acid were distilled off under reduced pressure. The residue was dissolved with THF (5 mL) and a functionalized aryl magnesium bromide (1 M in THF, 6.0 mL, 6.0 mmol) was added at 0 °C. The mixture was stirred for 2 h. Extractive workup and purification by silica gel column chromatography yielded the corresponding aryl-(3-butenyl)dimethyltin (3–4 mmol). The functionalized aryl-3-butenyldimethyltin compound could also be prepared by the addition of 3-butenylmagnesium bromide to aryldimethyltin trifluoroacetate, which was obtained by the reaction of diaryldimethyltin with CF₃CO₂H. Diaryldimethyltin was easily obtained from dichlorodimethyltin with two equivalents of ArMgBr. Trifluoromethylphenylmagnesium bromide and methoxyphenylmagnesium bromide were prepared from the corresponding aryl bromide and magnesium in THF according to the usual method. *tert*-Butoxycarbonylphenylmagnesium bromide was prepared in situ by treating *tert*-butyl iodobenzoate with an equimolar amount of isopropyl magnesium bromide in THF at –40 °C according to the literature.¹

Preparation of the Starting Aryl-3-iodoalkyldimethyltin

The preparation of **1b** is representative. Triethylborane (1.0 mL, 1.0 M hexane solution, 1.0 mmol) was added to a solution of 3-butenyldimethylphenyltin (2.81 g, 10 mmol) and perfluorobutyl iodide (*n*-C₄F₉I, 3.4 mL, 20 mmol) in benzene (10 mL) at 25 °C under air. A slightly exothermic

¹ Boymond, L.; Rottländer, M.; Cahiez, G.; Knochel, P. *Angew. Chem. Int. Ed. Engl.* **1998**, 37, 1701.

reaction started immediately. After the reaction mixture was stirred for 5 h, the solvent was evaporated. Purification of the residue by silica gel column chromatography provided **1b** (4.39 g, 7.0 mmol) in 70% yield.

A Typical Procedure for Aryl Migration from Tin to Carbon

The transformation of **1b** to **3b** is representative. Triethylborane (1.0 M hexane solution, 0.1 mL, 0.1 mmol) was added to a solution of the substrate **1b** (0.63 g, 1.0 mmol) in refluxing benzene (30 mL) under air. The resulting mixture was stirred for 2 h under reflux and was concentrated in vacuo. The crude product was dissolved in THF (2 mL) under an argon atmosphere and methylmagnesium iodide (1.0 M THF solution, 1.2 mL, 1.2 mmol) was added at 0 °C. After stirring for 30 min at room temperature, the reaction mixture was poured into water and extracted with ethyl acetate (10 mL \times 3). The combined organic layers were washed with brine, dried over Na₂SO₄, and concentrated in vacuo. The product was purified through a silica gel column to give **3b** (0.36 g, 0.70 mmol) in 70% yield.

(3-Iodo-5,5,6,6,7,7,8,8,8-nonafluorooctyl)dimethylphenyltin (1b): IR (neat) 2921, 1429, 1352, 1297, 1238, 1134, 879, 726, 699 cm⁻¹. ¹H NMR (CDCl₃) 0.33 (t, *J* = 27.0 Hz, 6H), 1.00-1.23 (m, 2H), 1.90-2.12 (m, 2H), 2.74-3.00 (m, 2H), 4.32 (m, 1H), 7.37-7.58 (m, 5H). ¹³C NMR (CDCl₃) -11.15 (t, *J* = 168 Hz), -11.12 (t, *J* = 168 Hz), 10.03 (t, *J* = 168 Hz), 25.63 (t, *J* = 39.8 Hz), 37.72, 40.38 (t, *J* = 20.8 Hz), 128.36 (t, *J* = 22.2 Hz), 128.62, 136.08 (t, *J* = 17.4 Hz), 140.68 (t, *J* = 232 Hz). ¹³C signals of the perfluorobutyl group were not observed. Anal. Calcd for C₁₆H₁₈F₉ISn: C, 30.65; H, 2.89. Found: C, 30.49; H, 2.93.

(3-Phenyl-5,5,6,6,7,7,8,8,8-nonafluorooctyl)trimethyltin (3b): IR (neat) 2914, 1355, 1296, 1133, 881, 759, 715 cm⁻¹. ¹H NMR (CDCl₃) 0.03 (t, *J* = 27.0 Hz, 9H), 0.51-0.70 (m, 2H), 1.72-1.96 (m, 2H), 2.30-2.48 (m, 2H), 2.95 (m, 1H), 7.15-7.35 (m, 5H). ¹³C NMR (CDCl₃) -10.74 (t, *J* = 162 Hz), 7.95 (t, *J* = 173 Hz), 34.21, 36.45 (t, *J* = 21.0 Hz), 42.12 (t, *J* = 29.6 Hz), 126.77, 127.62, 128.71, 143.88. Anal. Calcd for C₁₇H₂₁F₉Sn: C, 39.64; H, 4.11. Found: C, 39.81; H, 4.14.

(3-Iodo-5,5,6,6,7,7,8,8,9,9,10,10,11,11,12,12,12-heptafluorododecyl)(2-

methoxyphenyl)dimethyltin (8a): IR (neat) 2908, 1577, 1460, 1431, 1236, 1147 cm^{-1} ; ^1H NMR (CDCl_3) 0.28 (t, $J = 27.0$ Hz, 6H), 0.98-1.20 (m, 2H), 1.90-2.07 (m, 2H), 2.76-2.93 (m, 2H), 3.79 (s, 3H), 4.33 (m, 1H), 6.84 (d, $J = 8.1$ Hz, 1H), 7.00 (t, $J = 7.0$ Hz, 1H), 7.31-7.37 (m, 1H); ^{13}C NMR (CDCl_3) -10.65 (t, $J = 172$ Hz, 2C), 10.59 (t, $J = 175$ Hz), 26.32 (t, $J = 41.5$ Hz), 37.94, 40.42 (t, $J = 20.5$ Hz), 55.16, 109.20 (t, $J = 11.4$ Hz), 121.22 (t, $J = 22.8$ Hz), 129.00, 130.42, 136.61 (t, $J = 12.8$ Hz), 163.80; Anal. Calcd for $\text{C}_{21}\text{H}_{20}\text{F}_{17}\text{IOSn}$: C, 29.43; H, 2.35. Found: C, 29.72; H, 2.38.

(3-Iodo-5,5,6,6,7,7,8,8,9,9,10,10,11,11,12,12-heptafluorododecyl)(3-methoxyphenyl)dimethyltin (8b): IR (neat) 2908, 1585, 1476, 1411, 1210, 1042 cm^{-1} ; ^1H NMR (CDCl_3) 0.32 (t, $J = 26.5$ Hz, 6H), 1.01-1.19 (m, 2H), 1.93-2.07 (m, 2H), 2.76-2.95 (m, 2H), 3.82 (s, 3H), 4.32 (m, 1H), 6.87 (d, $J = 7.5$ Hz, 1H), 7.01 (s, 1H), 7.04 (d, $J = 7.2$ Hz, 1H), 7.30 (dd, $J = 7.5$ Hz, 7.2 Hz, 1H); ^{13}C NMR (CDCl_3) -10.90 (t, $J = 170$ Hz), -10.87 (t, $J = 170$ Hz), 10.09 (t, $J = 169$ Hz), 25.87 (t, $J = 40.4$ Hz), 37.67, 40.40 (t, $J = 20.5$ Hz), 55.01, 113.74, 121.57 (t, $J = 20.0$ Hz), 128.21 (t, $J = 15.0$ Hz), 129.25 (t, $J = 25.7$ Hz), 142.19 (t, $J = 227$ Hz), 159.33 (t, $J = 27.8$ Hz); Anal. Calcd for $\text{C}_{21}\text{H}_{20}\text{F}_{17}\text{IOSn}$: C, 29.43; H, 2.35. Found: C, 29.55; H, 2.33.

(3-Iodo-5,5,6,6,7,7,8,8,8-nonafluorooctyl)(2-trifluoromethylphenyl)dimethyltin (8d): IR (neat) 2916, 1438, 1237, 1122, 1031, 880, 771 cm^{-1} ; ^1H NMR (CDCl_3) 0.38 (t, $J = 27.0$ Hz, 6H), 1.06-1.15 (m, 2H), 1.85-2.04 (m, 2H), 2.74-2.93 (m, 2H), 4.31 (m, 1H), 7.42-7.52 (m, 2H), 7.61 (d, $J = 7.2$ Hz, 1H), 7.71 (d, $J = 7.5$ Hz, 1H); ^{13}C NMR (CDCl_3) -9.22 (t, $J = 171$ Hz), -9.15 (t, $J = 171$ Hz), 11.30 (t, $J = 172$ Hz), 25.48 (t, $J = 40.1$ Hz), 37.48, 40.35 (t, $J = 20.5$ Hz), 125.18 (q, $J = 272.0$ Hz), 126.04 (q, $J = 4.6$ Hz), 128.75, 130.98 (t, $J = 19.5$ Hz), 136.39 (t, $J = 30.2$ Hz), 137.15 (t, $J = 13.1$ Hz), 139.08; Anal. Calcd for $\text{C}_{21}\text{H}_{20}\text{F}_{17}\text{IOSn}$: C, 29.38; H, 2.47. Found: C, 29.65; H, 2.41.

(3-Iodo-5,5,6,6,7,7,8,8,8-nonafluorooctyl)(4-trifluoromethylphenyl)dimethyltin (8f): IR (neat) 1602, 1327, 1236, 1132, 1051, 879, 821, 738 cm^{-1} ; ^1H NMR (CDCl_3) 0.37 (t, $J = 26.7$ Hz, 6H), 1.04-1.26 (m, 2H), 1.90-2.06 (m, 2H), 2.69-3.00 (m, 2H), 4.31 (m, 1H), 7.50-7.66 (m, 4H); ^{13}C NMR (CDCl_3) -10.97 (t, $J = 171$ Hz), -10.93 (t, $J = 171$ Hz), 10.13 (t, $J = 172$ Hz),

25.55 (t, $J = 40.1$ Hz), 37.47, 40.41 (t, $J = 20.5$ Hz), 124.40 (q, $J = 271$ Hz), 124.59 (tq, $J = 22.2$ Hz, 3.5 Hz), 130.81 (q, $J = 32.5$ Hz), 136.30 (t, $J = 17.7$ Hz), 146.34; Anal. Calcd for $C_{21}H_{20}F_{17}IO_2Sn$: C, 29.38; H, 2.47. Found: C, 29.78; H, 2.36.

(3-Iodo-5,5,6,6,7,7,8,8,8-nonafluorooctyl)(3-*tert*-butoxycarbonylphenyl)dimethyltin (8h):

IR (neat) 2974, 1713, 1294, 1237, 1134, 742 cm^{-1} ; 1H NMR ($CDCl_3$) 0.36 (t, $J = 26.7$ Hz, 6H), 1.03-1.25 (m, 2H), 1.60 (s, 9H), 1.92-2.07 (m, 2H), 2.76-2.94 (m, 2H), 4.31 (m, 1H), 7.40 (dd, $J = 7.2$ Hz, 6.6 Hz, 1H), 7.62 (d, $J = 7.2$ Hz, 1H), 7.93 (d, $J = 6.6$ Hz, 1H), 8.10 (s, 1H); ^{13}C NMR ($CDCl_3$) -10.84 (t, $J = 172$ Hz), -10.81 (t, $J = 172$ Hz), 10.14 (t, $J = 171$ Hz), 25.62 (t, $J = 41.0$ Hz), 28.07, 37.57, 40.31 (t, $J = 20.5$ Hz), 81.00, 127.96 (t, $J = 22.5$ Hz), 129.61, 131.51 (t, $J = 20.5$ Hz), 136.76 (t, $J = 18.8$ Hz), 140.02 (t, $J = 17.7$ Hz), 140.95, 166.22; HRMS (m/z): calcd for $C_{20}H_{23}F_9IO_2^{120}Sn$ (M^+-Me), 712.9583; found, 712.9625.

(3-Iodo-5,5,6,6,7,7,8,8,8-nonafluorooctyl)(4-*tert*-butoxycarbonylphenyl)dimethyltin (8i):

IR (neat) 2972, 1711, 1589, 1294, 1237, 1135, 750 cm^{-1} ; 1H NMR ($CDCl_3$) 0.35 (t, $J = 26.7$ Hz, 6H), 1.04-1.24 (m, 2H), 1.59 (s, 9H), 1.89-2.08 (m, 2H), 2.74-2.96 (m, 2H), 4.31 (m, 1H), 7.53 (d, $J = 8.2$ Hz, 2H), 7.94 (d, $J = 8.2$ Hz, 2H); ^{13}C NMR ($CDCl_3$) -10.87 (t, $J = 172$ Hz), -10.83 (t, $J = 172$ Hz), 10.12 (t, $J = 171$ Hz), 25.61 (t, $J = 41.0$ Hz), 28.08, 37.52, 40.31 (t, $J = 20.5$ Hz), 80.92, 128.67 (t, $J = 22.2$ Hz), 132.15, 135.85 (t, $J = 17.6$ Hz), 147.16, 166.10; HRMS (m/z): calcd for Calcd for $C_{20}H_{23}F_9IO_2^{120}Sn$ (M^+-Me), 712.9583; found, 712.9625.